Answer For Basic Stoichiometry Pogil Activity

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Answer For Basic Stoichiometry Pogil

HS Chemistry POGIL Activity Page 3 Basic Stoichiometry Model 3 Given the following equation: _____ N 2 (g) + _____ H 2 (g) → _____ NH 3 (g) 9. If 0.052 mol N 2 are reacted, how many mol NH 3 are formed? Using dimensional analysis, show how you

calculated your answer. 10.

HS Chemistry POGIL Activity Basic Stoichiometry

HS Chemistry POGIL Activity Page 2. Basic Stoichiometry. Molar Equivalence. Gas volume. of A at STP (in Liters) Number of. Particles . of A. Mole A. Mass of A (in grams) 1 mole. GFW. GFW. 1mole. 1 mole. 6.023 x1023. 6.023 x 1023. 1

mole. 22.4 L. 1 mole. 1 mole. 22.4 L. Mole B. Number of. Particles . of B. 1 mole. 6.023 x1023. 6.023 x 1023. 1 mole. 22.4 L. 1 mole. 1 mole. 22.4 L. Gas volume. of B at STP (in Liters)

HS Chemistry POGIL Activity

HS Chemistry POGIL Activity Page 5 Basic Stoichiometry _____ H 2 (g) + _____ O 2 (g) _____ H 2 O (g) 14.

Given the equation above, determine the number of moles of water produced when 5.2 g O 2 are reacted. Make sure to show the dimensional analysis in your work.

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MODEL 1: Mole-to-Mole Stoichiometry (1-step) Example: mole-to-mole Question: Given the following equation:

wod be 1N2 (g)+3 H2 (g) 2 NH, (g) How many moles of H2 are necded to produce 34.8 moles of NH,? Mole ratio Mole ratio Answer: 34.8 mol NH3 x 3 mol H 52,2mol H2 2 mol NHy 34.8 mol NH3 mol H2 52.2mol H 2 mol NH3 OR Key Questions: 1.

Solved: POGIL- Stoichiometry How Do Chemists Use Balanced C ...

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HS Chemistry POGIL

Activity Topic: Naming & Formula Writing 1(BW) Particle connections – What's in a name? Why? In this activity we will address the question: How do the smallest particles of matter connect to . each other and how do we represent those connections by the names we give a substance? Figure 1 . Particle . model

Chemistry POGIL

Activity «Activity

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