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Chapter 10 Answers

Solutions Manual Chemistry: Matter and Change • Chapter 10 163 CHAPTER 10 SOLUTIONS MANUAL 3. Apply How many atoms of hydrogen-1 are in a 1.007-g sample? Recall that 1.007 amu is the mass of one atom of hydrogen-1. Round your answer to two significant digits.

1.007 g $1 \text{ H-1 atom} \times 1.672 \times 10^{-24} \text{ g a.}$

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6.023 1023 H-1 atoms 4.

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Matching Match each item with the correct statement below.

a. calorimeter
d. temperature
b. calorie
e. specific heat
c. joule
f. heat

___ 1. quantity of heat needed to raise the temperature of 1 g

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of water by 1 C ____ 2. SI unit of energy
____ 3. energy transferred between 2
objects because of temperature
difference

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SOLUTIONS MANUAL Section 11.1

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Assessment page 372 5. Compare the mass of the reactants and the mass of the products in a chemical reaction, and explain how these masses are related. The coefficients in the balanced equation indicate the molar relationship between each ...

StoichiometryStoichiometry

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Solutions to In-Chapter Problems

10.1 Refer to Example 10.1 to answer the question. • The atomic number (Z) = the number of protons. • The mass number (A) = the number of protons + the number of neutrons.

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Modern Chemistry 86 Chapter Test

Name Class Date Chapter Test A,

continued _____ 13. The separation

process of paper chromatography can

be explained by a. vaporization of the

ink. b. water vapor pressure. c. capillary

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action. d. pull of gravity. _____14. When there is a small decrease in temperature, the average kinetic energy of the particles ...

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16. The measure of the ability of an
atom in a chemical compound to attract
electrons from another atom in the

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compound is called _____.

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